

Rešitve

- |     |   |     |
|-----|---|-----|
| 1.1 | D | 3 T |
| 1.2 | C | 3 T |
| 1.3 | E | 3 T |

Skupaj: 9,0 T

2.

	berilijev diklorid	ogljikov dioksid	vodikov cianid
strukturna formula molekule	 $1,0\ T$	 $1,0\ T$	$\text{H}-\text{C}\equiv\text{N}:$ $1,0\ T$
polarnost molekule	nepolarna $0,5\ T$	nepolarna $0,5\ T$	polarna $0,5\ T$

Skupaj: 9,0 T

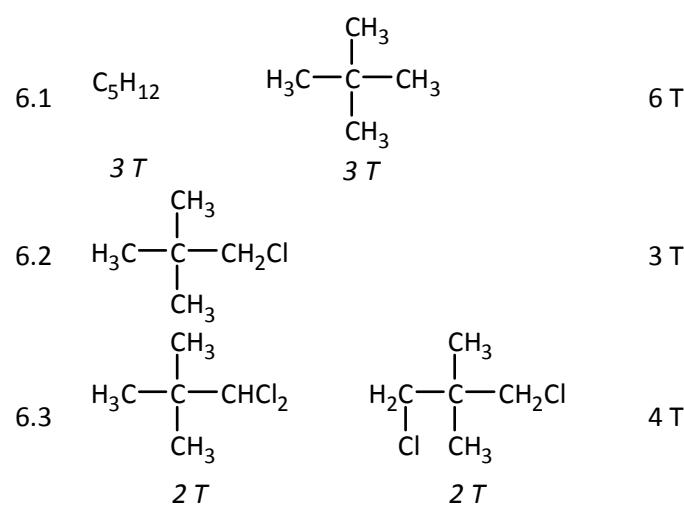
- 3.1  $-2219 \text{ kJ}$       4 T  
 3.2  $-2,29 \cdot 10^4 \text{ kJ}$  (ali sprosti se  $2,29 \cdot 10^4 \text{ kJ}$  ali  $2,29 \cdot 10^4 \text{ kJ}$ ) 4 T

Skupaj: 8,0 T

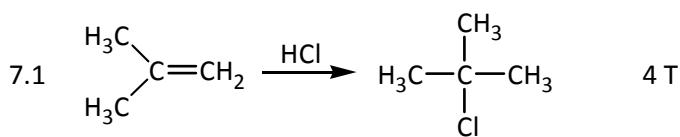
- $$4. \quad 7.8 \cdot 10^{23} \text{ ionov}$$

- |    |   |                                  |             |
|----|---|----------------------------------|-------------|
| 5. | A | $\text{NH}_3(\text{g})$          | $4\text{T}$ |
|    | B | $\text{NH}_4\text{Cl}(\text{s})$ | $4\text{T}$ |

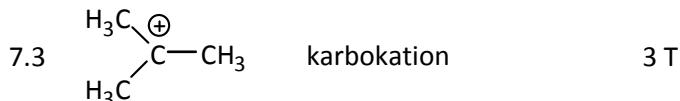
Skupaj: 8,0 T



Skupaj: 13,0 T

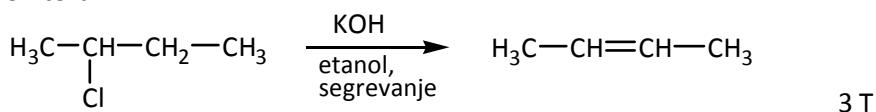


- ## 7.2 elektrofilna adicija 2 T

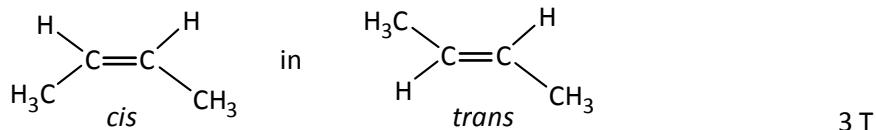


Skupaj: 9,0 T

## 8.1 Sinteza:



Problem te sinteze je nastanek dveh geometrijskih izomerov.



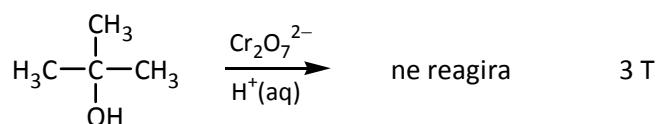
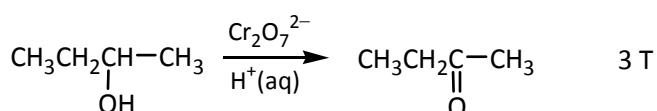
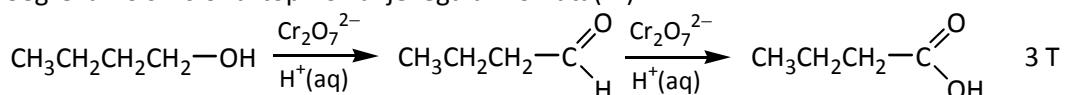
## 8.2



**Skupaj: 10,0 T**

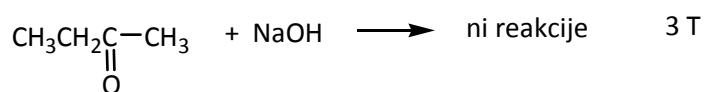
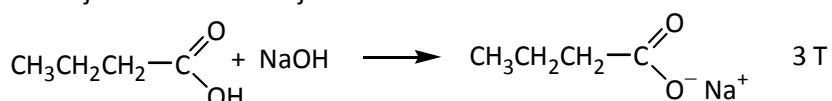
## 9.1 1. test

Segrevamo s kislo raztopino kalijevega dikromata(VI).



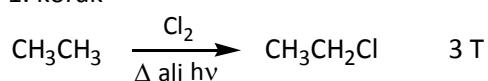
## 9.2 2. test

Reakcija z NaOH ali natrijem

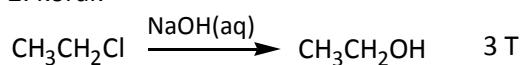


**Skupaj: 15,0 T**

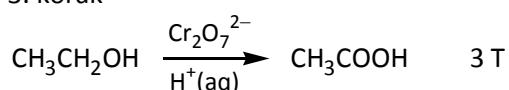
10.1 1. korak



10.2 2. korak



10.3 3. korak



10.4 4. korak

**Skupaj: 12,0 T****Vse skupaj: 100,0 T**